

Course: Chem 506 (Spring 2017)

Topic: Quantum Chemistry & Spectroscopy

Instructors: Paras Prasad, Alexander Baev

Contact: Email: pnprasad@buffalo.edu; abaev@buffalo.edu

Phone: 716 645-4148

OFFICE Hours: by appt only (Contact my assistant Margie Weber, mdweber@buffalo.edu)

ROOM: 508 Cooke

TIME: Tuesday, Thursday, 11:00AM-12:20PM

CONTENTS*

Part I

Schedule

- | | |
|--|------------------------------------|
| <p>1. Quantum Mechanics (QM) vs Classical mechanics.
Postulates of QM. Application to chemical atomic systems:
Schrodinger equation for hydrogen atom
and its solution. Quantum numbers. Selection rules.
Spin. Uncertainty principle.</p> | <p>January 31, Feb 2 (Prasad)</p> |
| <p>2. Many electron atoms and electron-electron interaction;
Pauli's principle.
Hartree-Fock method for many electron systems.
Electronic configuration of many-electron atoms.
Popular Quantum Chemistry codes. Semi-empirical methods.</p> | <p>February 7, 9 (Prasad)</p> |
| <p>3. Approximate methods for complex chemical systems.
Variation and perturbation
treatments; examples of each calculations of first
and second order perturbation terms.</p> | <p>February 14, 16 (Baev)</p> |
| <p>4. Molecular systems: partitioning of
molecular Hamiltonian into electronic, vibrational
and rotational terms by using Born-Oppenheimer
approximation.</p> | <p>February 21, 23 (Baev)</p> |
| <p>5. Valence Bond (V.B) vs. Molecular orbitals (MO):
Homo-nuclear and heteronuclear diatomic molecules.
More sophisticated molecules and
symmetry requirements.</p> | <p>February 28, March 2 (Baev)</p> |

EXAM I

March 7

Part II

- | | |
|---|---------------------------|
| <p>1. A brief introduction to symmetry and group theory:
symmetry elements; point groups representations;
character tables. Applications of group theory.</p> | <p>March 9, 14 (Baev)</p> |
|---|---------------------------|

- | | |
|--|------------------------------------|
| <p>2. Brief discussions of following topics:
Ligand field theory.
Electronics structure of electron-deficient molecules
π-electron theory of organic molecules, Huckel's treatment
More sophisticated MO treatments</p> | <p>March 16, March 28 (Prasad)</p> |
| <p>3. Density Functional Theory for atomic and Molecular Clusters.</p> | <p>March 30, April 4 (Baev)</p> |
| <p>4. Nanomaterials; Quantum Dots, Metallic Nanoparticles and Plasmonics</p> | <p>April 6, April 11 (Baev)</p> |

EXAM II

April 13

Part III

- | | |
|---|--|
| <p>1. Spectroscopy: confirmation of quantum mechanical predictions;
energy states and transitions; interaction between molecules and radiation,
dipole allowed transitions;
Raman transitions; population inversion and lasers.</p> | <p>April 18, 20 (Prasad)</p> |
| <p>2. Multi-photon processes.
Nonlinear optical effects and applications</p> | <p>April 25 (Prasad)</p> |
| <p>3. Electronic spectroscopy and structure</p> | <p>April 27, May 2 (Prasad, Baev)</p> |
| <p>4. Vibrational spectroscopy, IR and Raman</p> | <p>May 4, (Prasad) May 9,(Baev)
May 11, (Prasad)</p> |

EXAM III

May 16

- Each Exam is 1 hour and 20 minutes and only covers the new materials. Each Exam is also equally weighted (100 points each)

Text Book

- "Quantum Chemistry and Spectroscopy" T. Engel, Prentice Hall (2010)

Supplementary Book:

"Quantum Chemistry" I.N. Levine, Fifth Edition, Prentice Hall (2000)